

PFAS in soil - forever pollution, forever concern?
25 – 26 March 2025, Berlin

Remediation of PFAS in Soil and Groundwater: State of the Art.

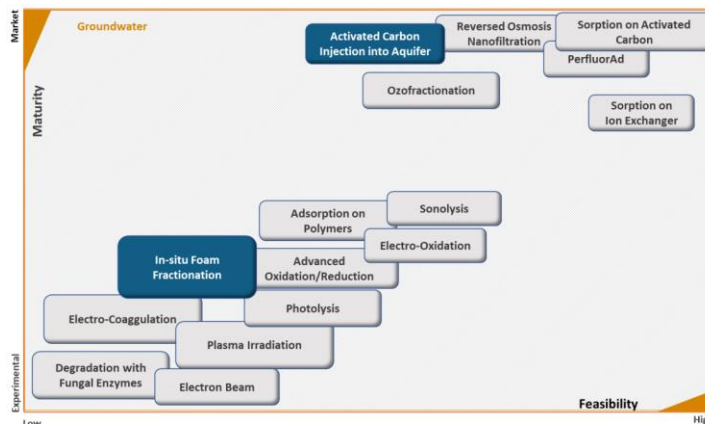
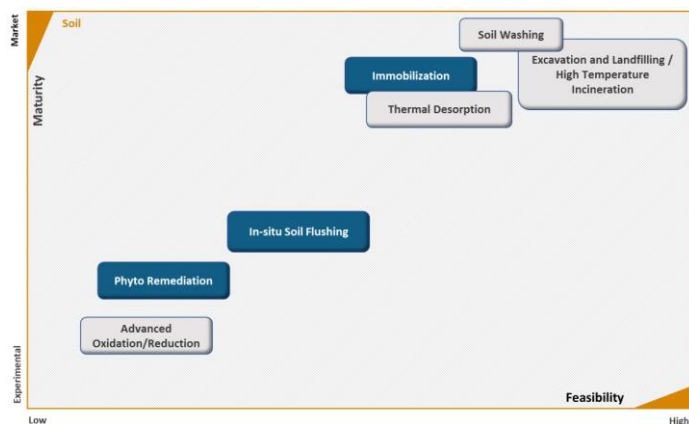
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Remediation Technologies – Soil and Groundwater

TEXTE
205/2020

Remediation management
for local and wide-spread
PFAS contaminations

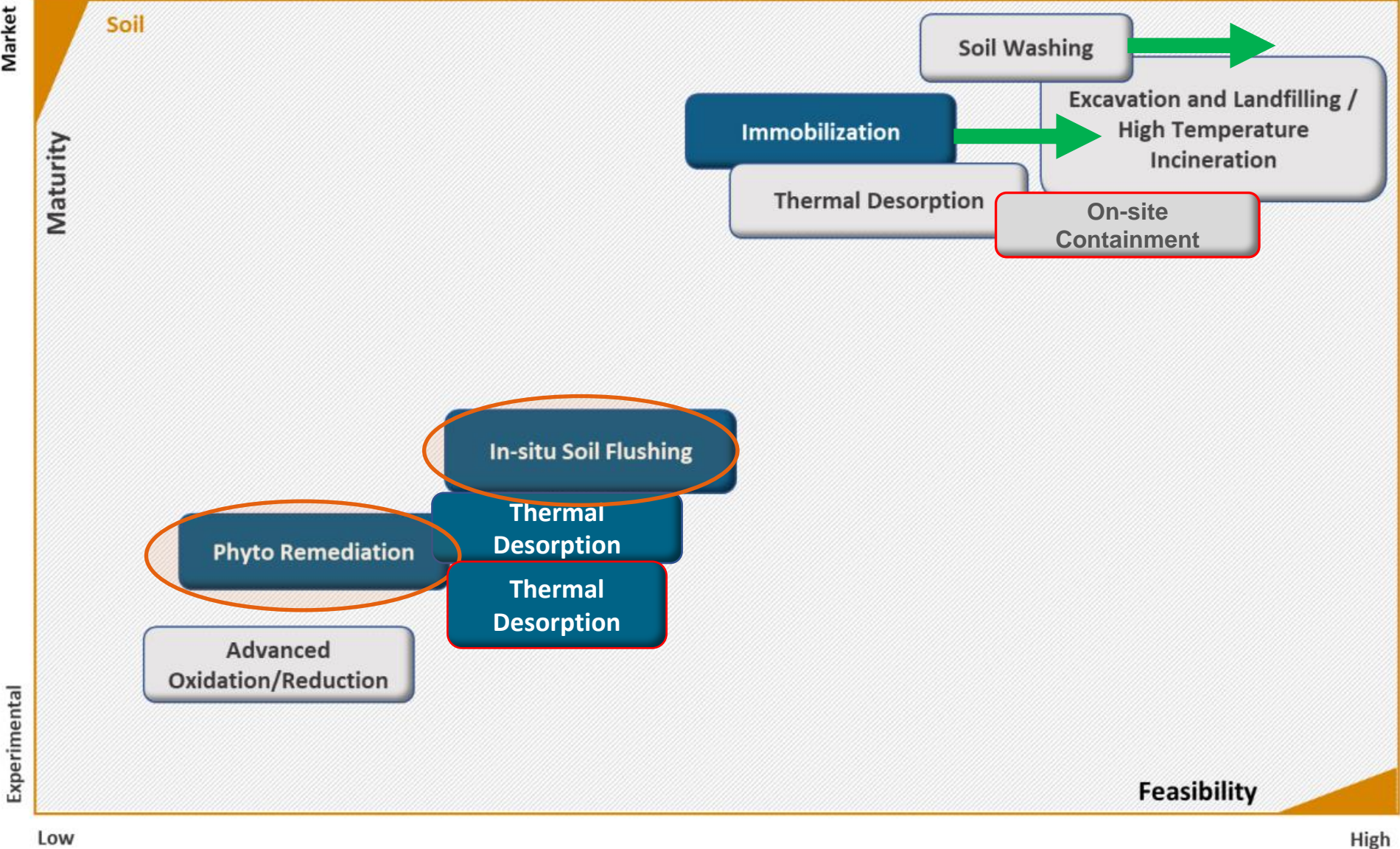


For our Environment

Umwelt
Bundesamt

Remediation Technologies - Soil

Technologies for Soil Remediation



In situ applicable

Additional investigations

Technologies for Soil Remediation

- Immobilization & Landfilling: cost-efficient compared to landfilling alone
- In-situ immobilization: successfully tested in a 2½ year field trial (McDonough et al., 2021)
- Thermal desorption (in situ thermal treatment; ISTT): vapor extraction, cooling, GAC (pilot scale successful)*
- Phytoremediation: R&D confirmed what we already knew (low accumulation ratio, low efficiency for long-chain PFAS)

*) Ref: https://xcdacademy.s3.amazonaws.com/battelle/2024_Chlorinated/E3_1235_731_Fitzgerald.pdf

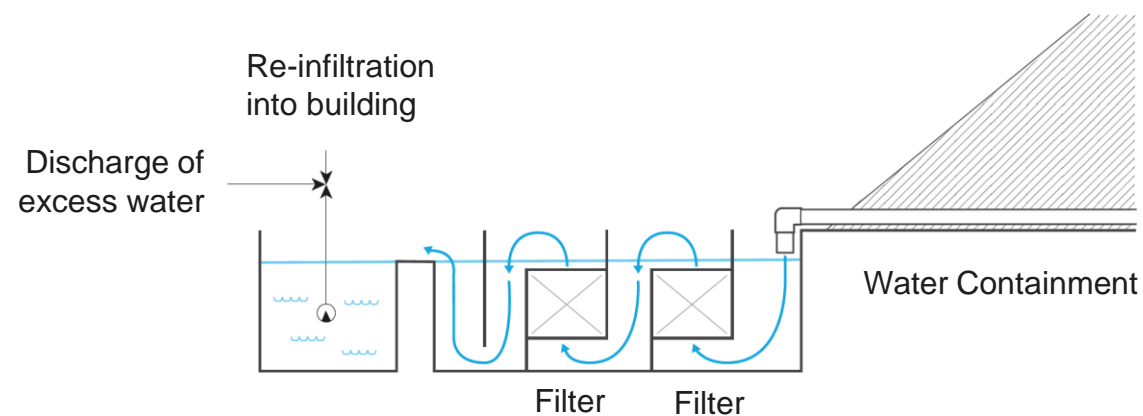
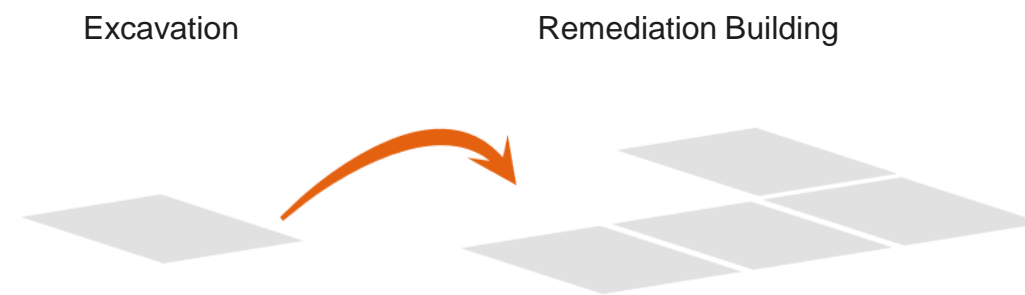
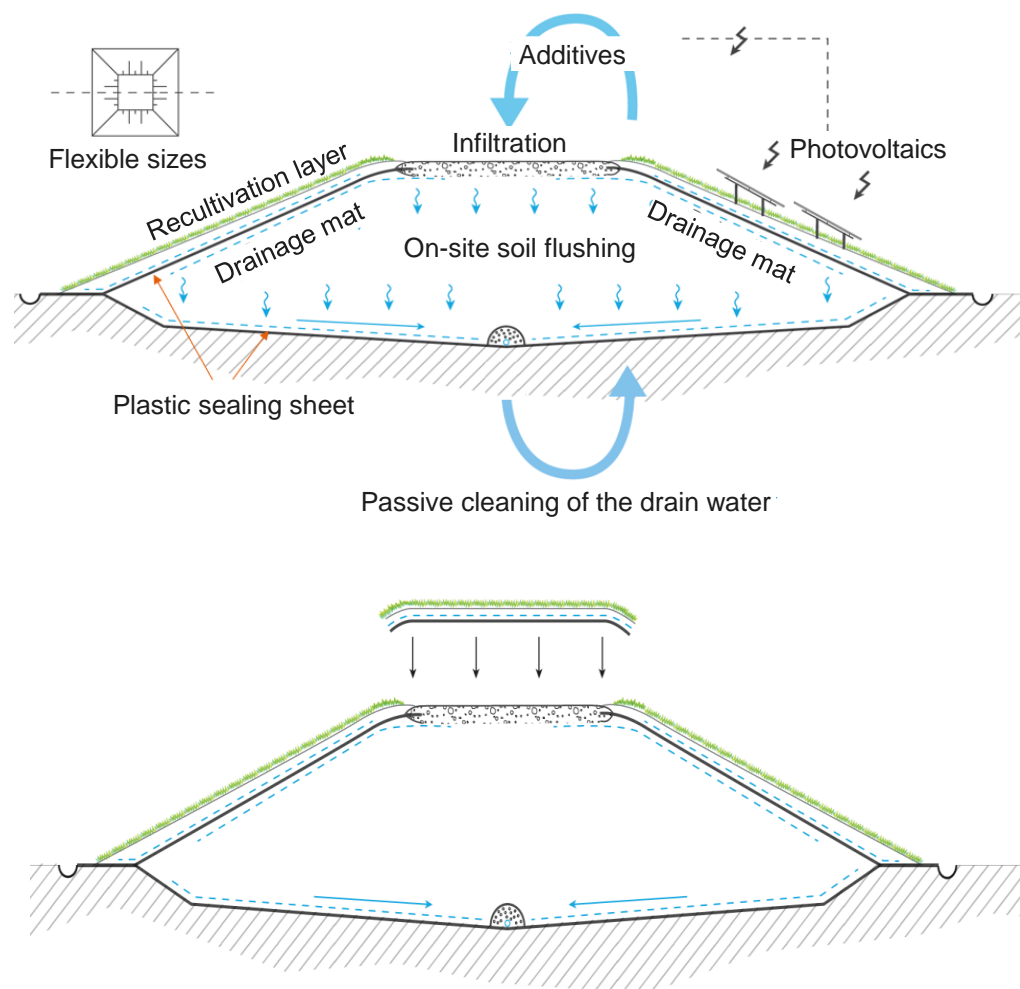
In-Situ Soil Flushing

- Only feasible for well permeable soils
- Still many issues unresolved:
 - *Achievability of target values with reasonable effort*
 - *Uniformity of water saturation (→ Non-Newtonian Fluids*) (BRGM: ethanol/xanthan gum)*
 - *Leachability in the dependence of the PFAS fingerprint (precursor, cationic, non-ionic PFAS)*
 - *Additional benefits of added (proteinic) biopolymers (→ water treatment)*
 - *Economic efficiency compared to dig & dump*

**NNF: viscosity changes depending on the acting shear forces. E.g. starch-water mixtures)

On-site Containment

Multifunctional remediation building

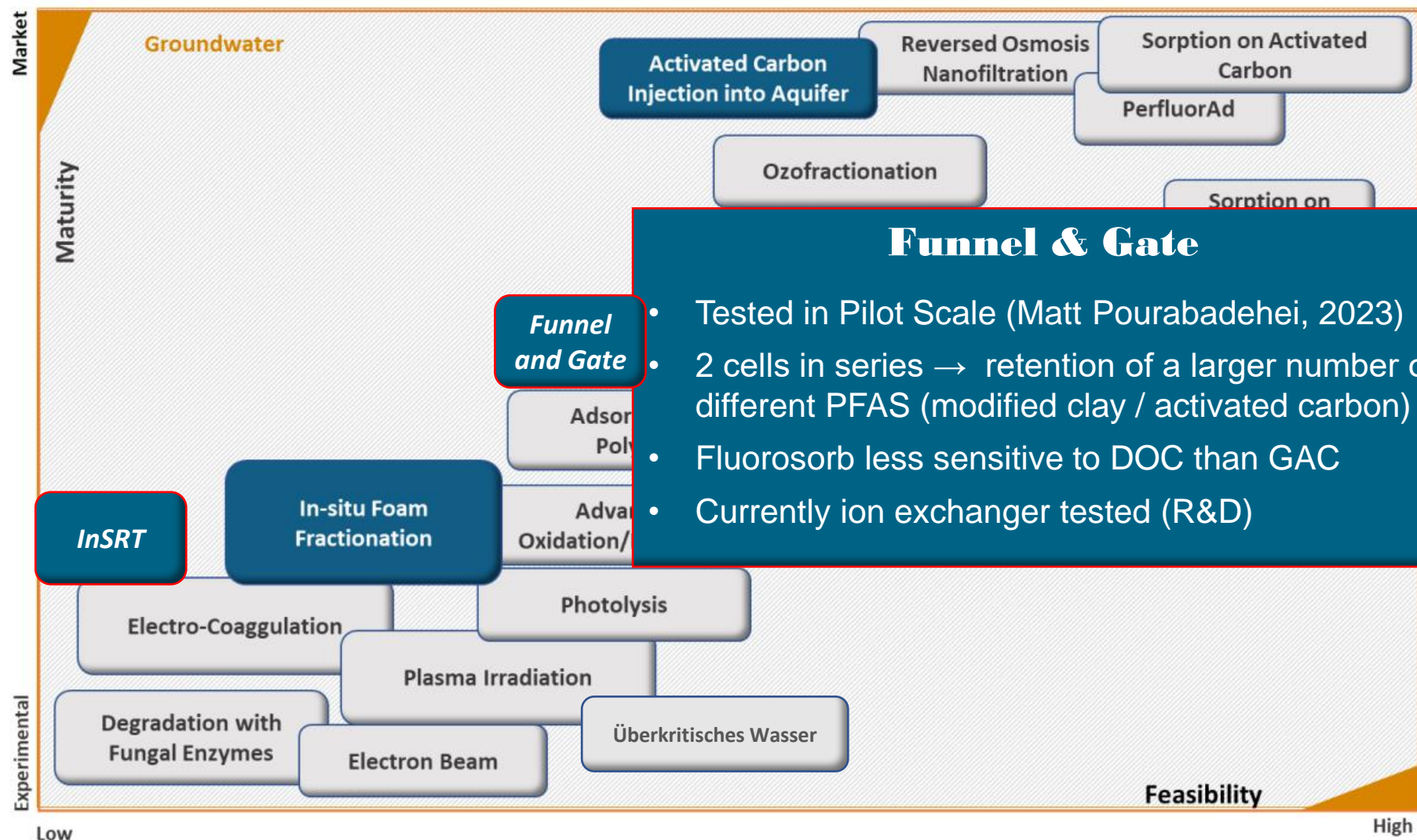


Remediation Technologies - Groundwater (in-situ)

Remediation Technologies for Groundwater

In-situ Technologies

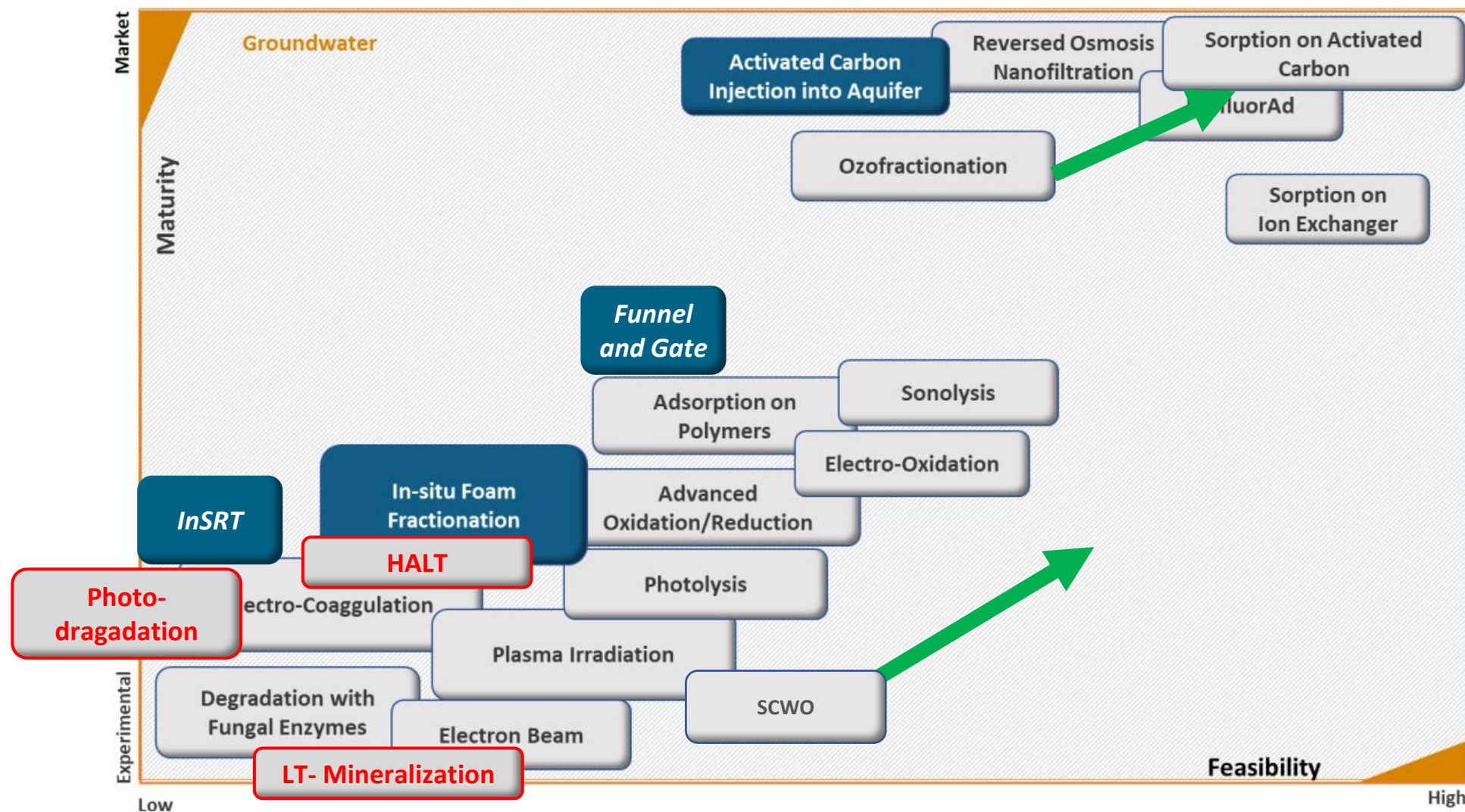
In situ applicable



InSRT
*In situ reactor
 technology systems
 based on HRX wells*

Remediation Technologies for Groundwater

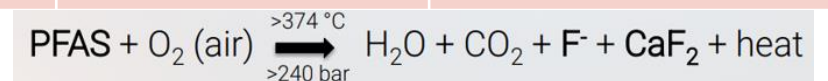
based on P&T



Ex-situ Groundwater Treatment

Closed F balance ?

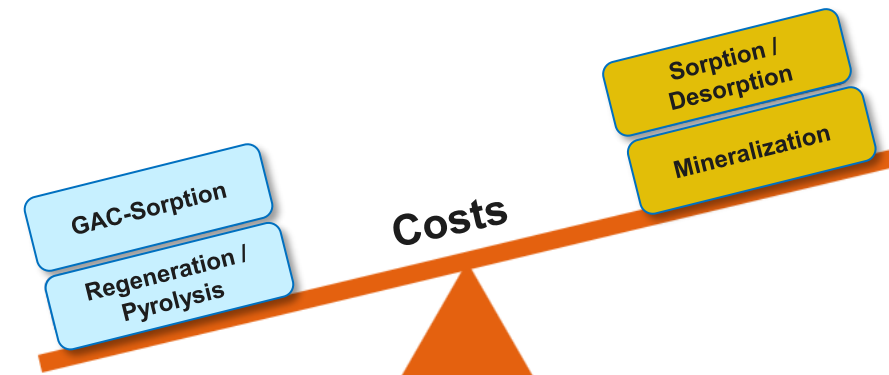
Technology	Environment	Degradation Efficiency	Pros /Cons	Source
Low Temperature Mineralization	DMSO/H ₂ O (8:1), NaOH, 120°C,	78 – 100 % degradation in 24 h	No degradation of PFSA	Trang et al., Science 377, 839–845 (2022)
Photodegradation	254 nm UV, 200 W Persulfat (S ₂ O ₈ ²⁻)	100 % PFOA-deg. in 4 h, PFOS poorly degraded		Verma, S., Next Materials 2 (2004) 100077
	UV/Sulfite/Iodide (20 mM SO ₃ ²⁻ , 10 mM I ⁻ , 10 mM HCO ₃ ⁻ , 150 mM OH ⁻)	100 % defluorination: PFOS (60 min), PFOA (20 min), PFBS (240 min)		O'Connor, N., Science of the Total Environment 888 (2023) 164137
HALT(Hydrothermal Alkaline Treatment)	250 bar, 350 °C, < 5 M NaOH	PFOS in 30 min zu 100 % abgebaut, PFAS _{gesamt} > 99,9 % Abbau	Solids also treatable	Gagliano, E. et al., Water Research 198, 2021, 117121
SCWO (Supercritical Water Oxidation)	> 374 °C, > 240 bar, O ₂ , Ca	PFAS _{total} > 99,9 % degradation in 30 sec.	Solids also treatable, acid formation and corrosion	EPA/600/R-22/257 September 2022



Ex-situ Groundwater Treatment

Interim Summary

- ❑ All (!) processes for the mineralization of PFAS require:
 - Harsh reaction conditions
 - High energy input
 - Mostly long treatment times
 - Expensive
- ❑ Unsuitable as a direct replacement for GAC sorption
- ❑ Concentrating up in small volumes (reversible sorption and desorption) required



Concentration Processes

PFAS Concentration Processes

Precursors & cationic
PFAS considered ?

Technology	Process	Status	Pros /Cons	Source
Electro sorption and desorption	Anodic potential → sorption, Reversing polarity → desorption (concentrate formation)	Pilot scale	Conductive GAC required	Georgi u. Mackenzie 28. Jahrg. 2022/ Nr.2, 53-57
Desorption using microwaves	Generates > 600 °C → > 90 % PFAS removal, weight loss < 5 % (3 min)	Lab scale	Regeneration efficiency decreases with increasing number of cycles, no formation of short-chain PFAS	Gagliano et al., Water Res. 198, 2021, 117121
Hydrothermal treatment	200 - 260 °C, high pressure, aqueous suspension, 4 h → PFOA degradation > 99 %	Lab scale	No formation of short-chain compounds .PFOS requires harsher reaction conditions (260 °C, 16 h)	
Regenerable magnetic anion ion exchangers (MIEX)	electrostatic sorption dominating over hydrophobic interactions	Pilot Scale	Fast (< 30 sec,) low efficiency for short chain PFAS, MeOH/salt for MIEX regen.	Tan, X. et al., Angew. Chem. Int. Ed. 2022, 61, e202213071
Foam fractionation	Injection of air in a water column	Full scale	Low efficiency for short chain PFAS	

Take Home Message

- Soil washing and immobilization (optional: in-situ) are becoming more and more established
- There are several processes available for the mineralization of PFAS in aqueous concentrates
- Bottleneck is still the concentration step
- New focus on in-situ barrier processes (especially sorption in funnel-and-gate systems)
- More R&D required in cost effective concentration processes
- **Economic feasibility studies are almost always missing**

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