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Grundlagen und Anwendungen der Physiologie-basierten Toxikokinetik-Modellierung



47. Sitzung der HBM Kommission
24./25. März 2014, Berlin

Outline

1. The potential
2. The structure of physiologically-based toxicokinetic models
3. Parameterization (incl. underlying model assumptions) and parameter sources
4. Example
5. Summary & References

The use of the term model in different contexts

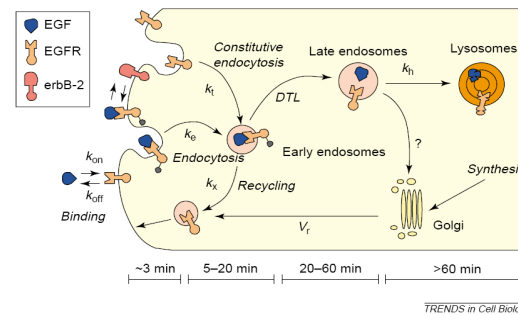
- **Model organisms**

- mouse as a model for human
- yeast as a simple eukaryote



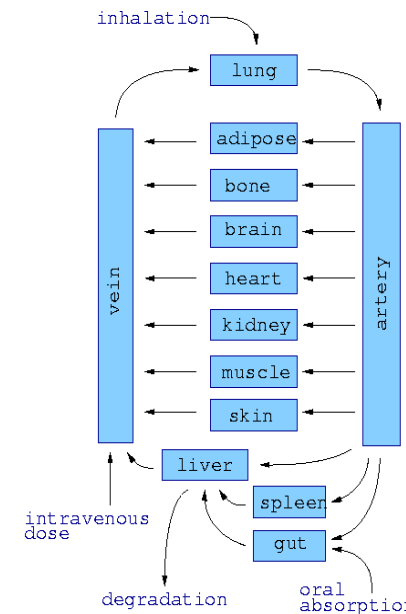
- **Biological model:**

- interaction network
- cartoon



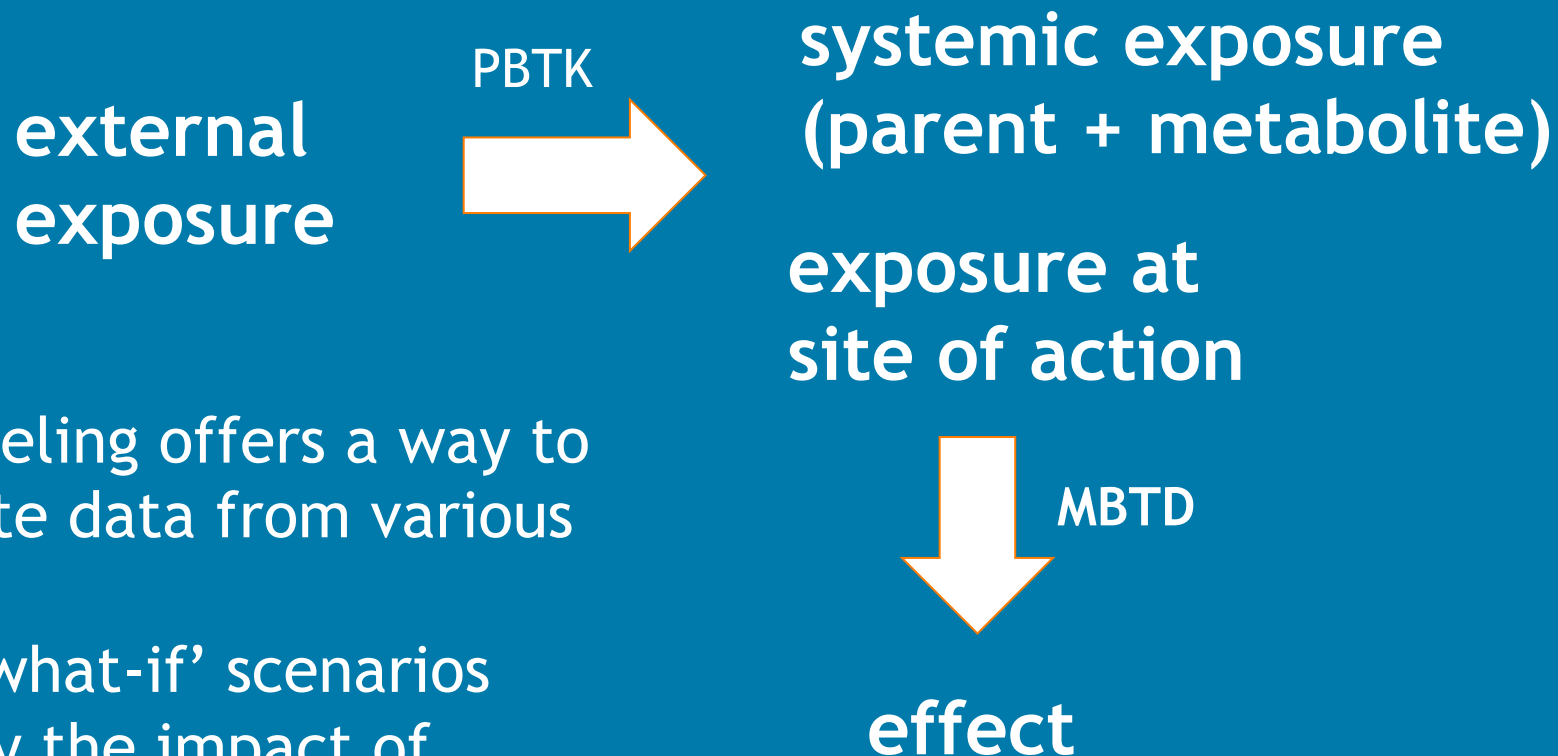
- **Mathematical model:**

- representation of biological model in terms of mathematical language
- in our context deterministic differential or algebraic equations



$$V_{org} \frac{d}{dt} C_{org} = Q_{org} \cdot \left(C_{in} - \frac{C_{org}}{P^{t:p}} \right) - CL \cdot C_{org}$$

The potential

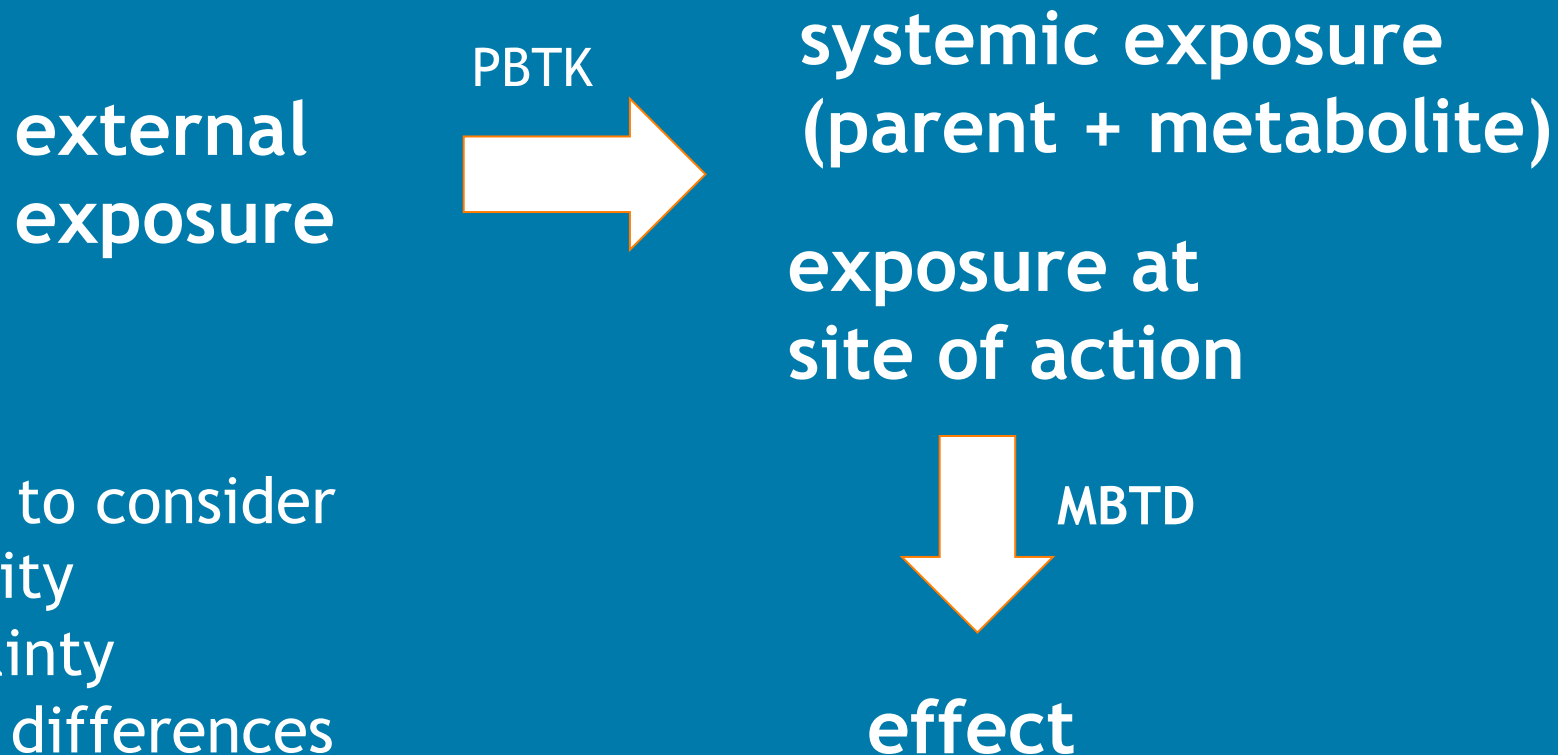


PBTk modeling offers a way to

- integrate data from various sources
- study 'what-if' scenarios
- quantify the impact of variability and uncertainty
- identify critical parameters
- ...

(PBTk= physiologically-based toxicokinetics, MBTD = mechanism-based toxicodynamics)

The potential



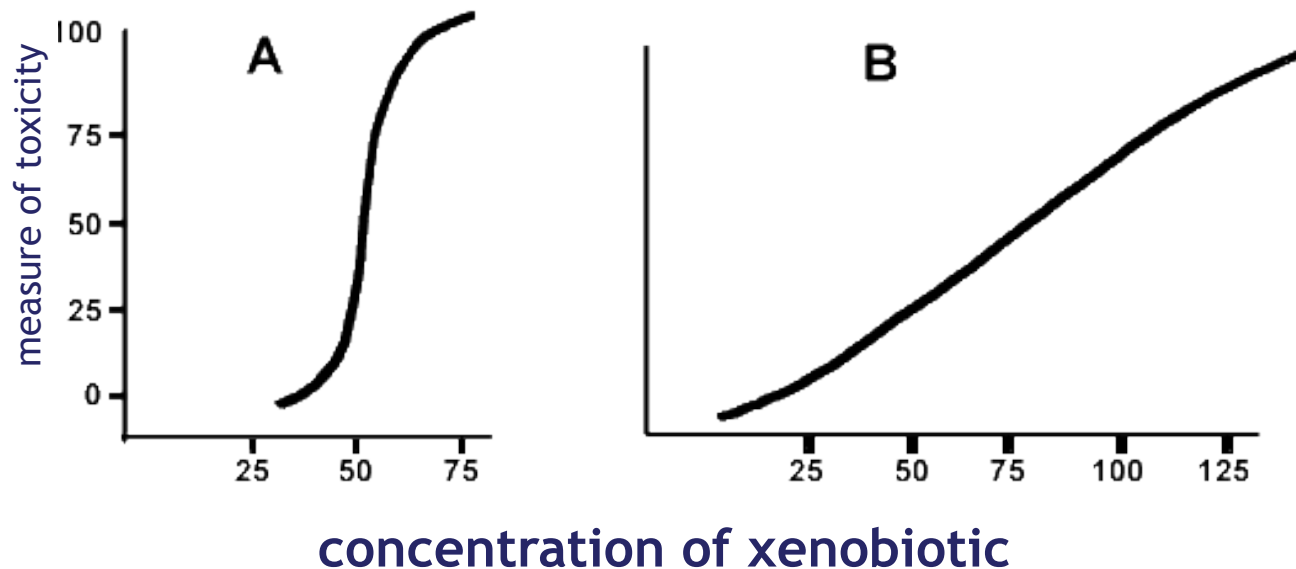
Important to consider

- variability
- uncertainty
- species differences
- in vitro-in vivo differences

Most critical bottleneck for application: availability of parameter values

Toxicity measured in in vitro assays

- Cell-culture in vitro assay data



→ Conclusions for toxicological risk assessment?

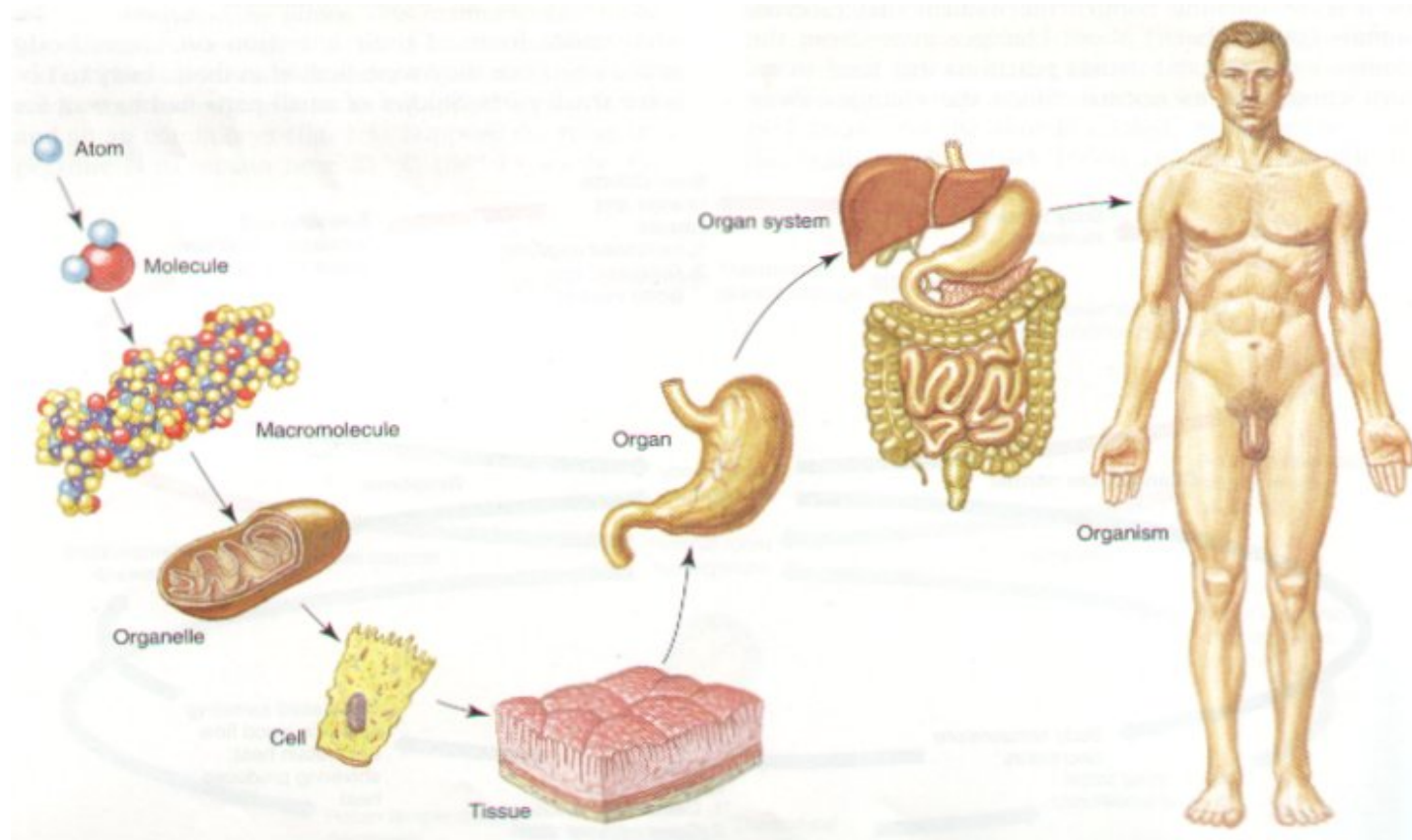
- Link from external exposure to systemic concentration and/or concentration at target site?
- Measure of exposure: C_{max} , AUC, ...? Parent compound or metabolite?

The structure of physiologically-based toxicokinetic models

PBTK modelling

- Main features:
 - mechanistic model of principal **ADME processes**:
(**A**bsorption, **D**istribution, **M**etabolism, and **E**xcretion)
 - compartments have anatomical interpretation
 - parameterized by physiological, anatomical and compound-specific data
- Long history in toxicokinetics
 - Teorell (1937) first PBPK model of therapeutically important, non-volatile chemical
 - Review: Gerlowski & Jain, *PBPK modelling: Principles and applications*, J Pharm Sci 72 (10), 1983
- More recent history in drug discovery and development
 - predominantly since 2000 due to large amount of parameters needed
 - breakthrough with seminal papers by Poulin&Theil, J Pharm Science (2000-2002)

Biological complexity



<http://www.sirinet.net/~jgjohnson/intro.html>

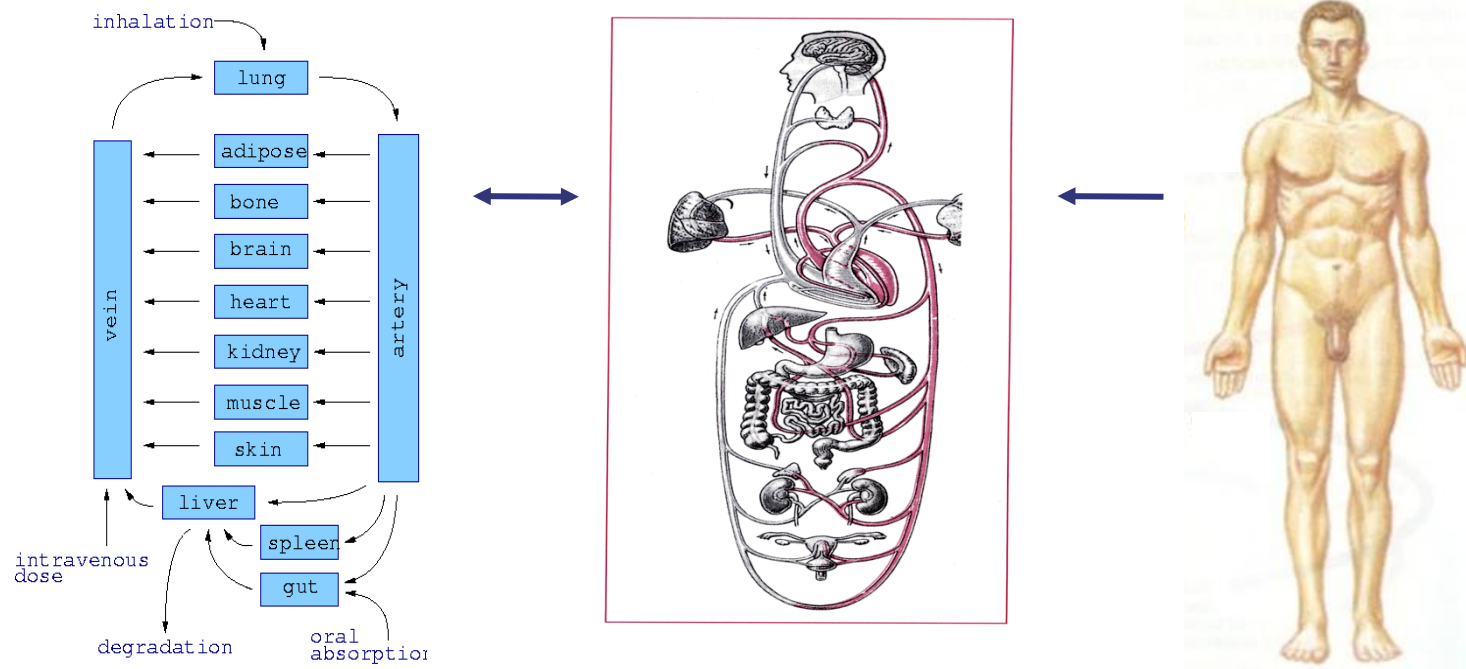
**Molecular
dynamics**

**Systems
biology**

Toxicokinetics

Physiologically based pharmacokinetic models

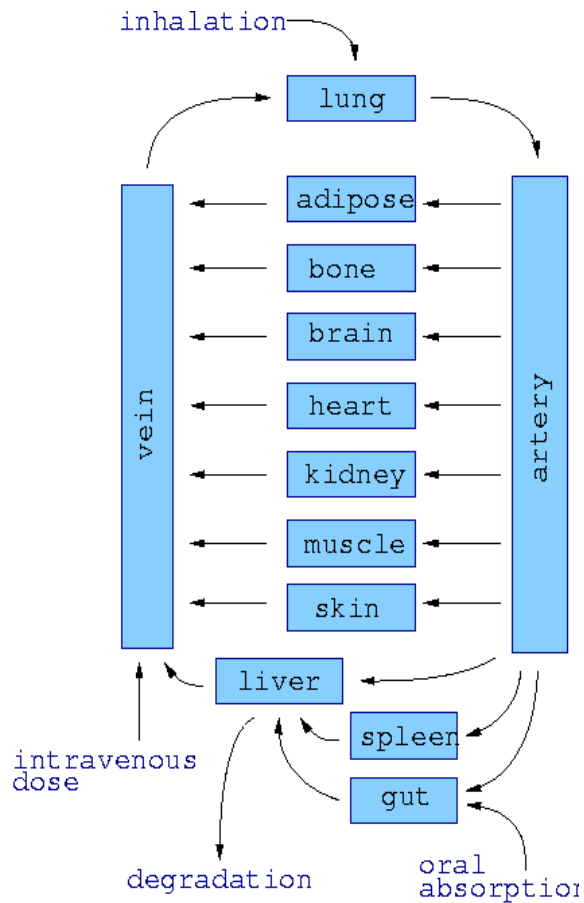
- Top down approach
 - organs, tissue and other spaces, interconnected by the blood flow



- Mass balance equations (ODEs) for concentrations in tissues/organs

Poulin & Theil, *J Pharm Sci.* (2002); Luepfert & Reichel, *Chem Biodiv.* (2005);
 Von Kleist, & Huisinga, *J Pharmacokinet Pharmacodyn* (2007)

Mass balance differential equations for each tissues/organ



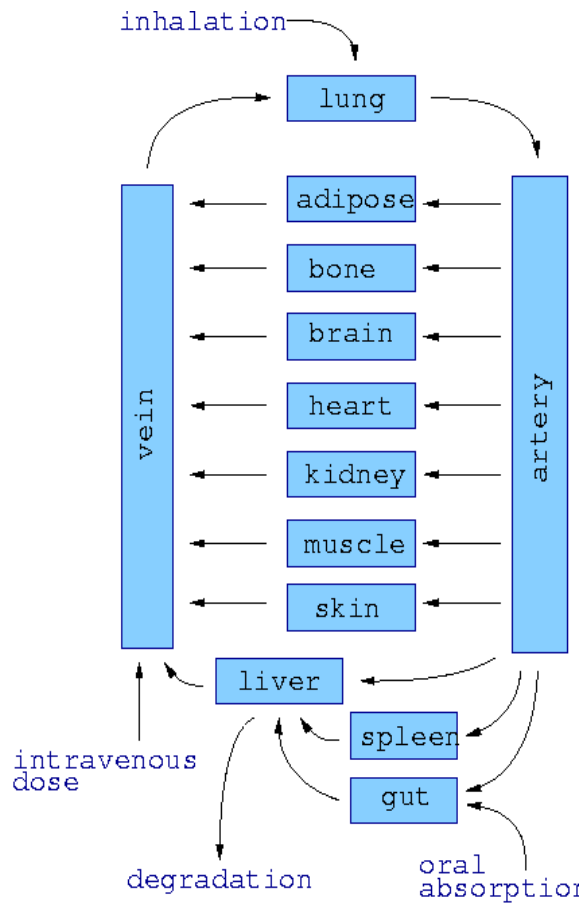
Well-stirred tissue model:

$$V_{tis} \frac{d}{dt} C_{tis} = Q_{tis} \cdot \left(\underset{\substack{\uparrow \\ \text{inflowing} \\ \text{blood conc.}}}{C_{in}} - \frac{C_{tis}}{\underset{\substack{\uparrow \\ \text{out-flowing} \\ \text{blood conc.}}}{K_{tis}}} \right) - CL_{int} \cdot C_{tis}$$

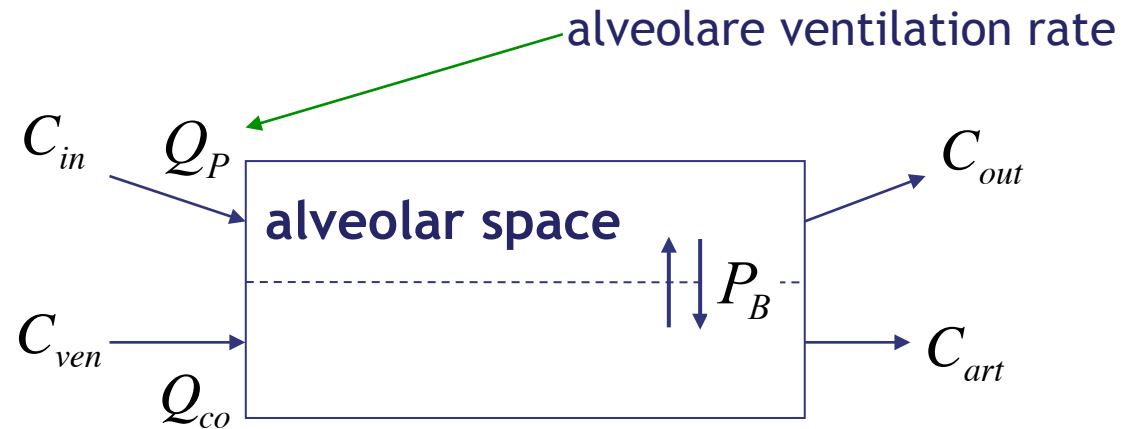
Steady-state tissue-to-blood partition coefficient:

$$K_{tis} = \frac{C_{tis,ss}}{C_{blood,ss}}$$

Mass balance differential equations for lung



Rate of change in lung:



Blood:air partition coefficient

$$P_{blo:air} = \frac{C_{art}}{C_{out}}$$

Poulin & Krishnan, Toxicol Appl Pharmacol (1996)

Parameterization and parameters sources

A rich source
of data
(human: adult
and children)

Annals of the ICRP

ICRP PUBLICATION 89

Basic Anatomical and Physiological Data
for Use in Radiological Protection:
Reference Values

Editor
J. VALENTIN

PUBLISHED FOR
The International Commission on Radiological Protection

By



PERGAMON

Accounting for inter-individual variability

Citation: *CPT: Pharmacometrics & Systems Pharmacology* (2012) 1, e4;
© 2012 ASCPT All rights reserved 2163-8306/12

www.nature.com/psp

ORIGINAL ARTICLE

Modeling Interindividual Variability in Physiologically Based Pharmacokinetics and Its Link to Mechanistic Covariate Modeling

W Huisinga¹, A Solms^{1,2}, L Fronton^{1,2} and S Pilari^{2,3,4}

**A rich source
of data
(animals:
mice and rats)**

**PHYSIOLOGICAL PARAMETER VALUES FOR
PHYSIOLOGICALLY BASED PHARMACOKINETIC MODELS**

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**¶National Center for Environmental Assessment
U.S. Environmental Protection Agency
Washington, DC**

Parameterization of PBPK models

Species specific

- blood flows, organ volumes

Drug specific

- intrinsic clearance (CL_{int})
- tissue-to-blood partition coefficients

- Administration (dose, route, etc)

Rate of change in tissue:

$$V_{tis} \frac{d}{dt} C_{tis} = Q_{tis} \cdot \left(\underset{\substack{\uparrow \\ \text{inflowing} \\ \text{blood conc.}}}{C_{in}} - \frac{C_{tis}}{\underset{\substack{\uparrow \\ \text{out-flowing} \\ \text{blood conc.}}}{K_{tis}}} \right) - CL_{int} \cdot C_{tis}$$

Tissue-to-blood partition coefficient:

$$K_{tis} = \frac{C_{tis,ss}}{C_{blood,ss}}$$

A-priori prediction of partition coefficients

- **Idea:** Consider tissue as composition of constituents important for xenobiotic distribution

tissue



Kandinskys Roten Fleck aufräumen

tissue constituents



water

neutral lipids

phospholipids

proteins

etc

“Kunst aufgeräumt“ by Ursus Wehrli

More mathematical

- Ansatz based on mass balance equation

interstitial water

binding proteins

cellular water

neutral lipids

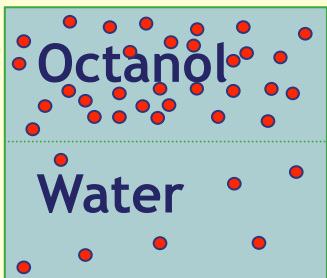
phospholipids

$$A_t = A_{wi} + A_{pr} + A_{wc} + A_{nl} + A_{ph}$$

$$= V^i (C_{ui} + C_{pr}) + V^{wc} C_{uc} + V^{nl} C_{nl} + V^{ph} C_{ph}$$

$$\frac{C_t}{C_{up}} = V^{i:t} \left(1 + \frac{C_{pr}}{C_{up}} \right) + V^{wc:t} \left(1 + V^{nl:t} \frac{C_{nl}}{C_{up}} + V^{ph:t} \frac{C_{ph}}{C_{up}} \right)$$

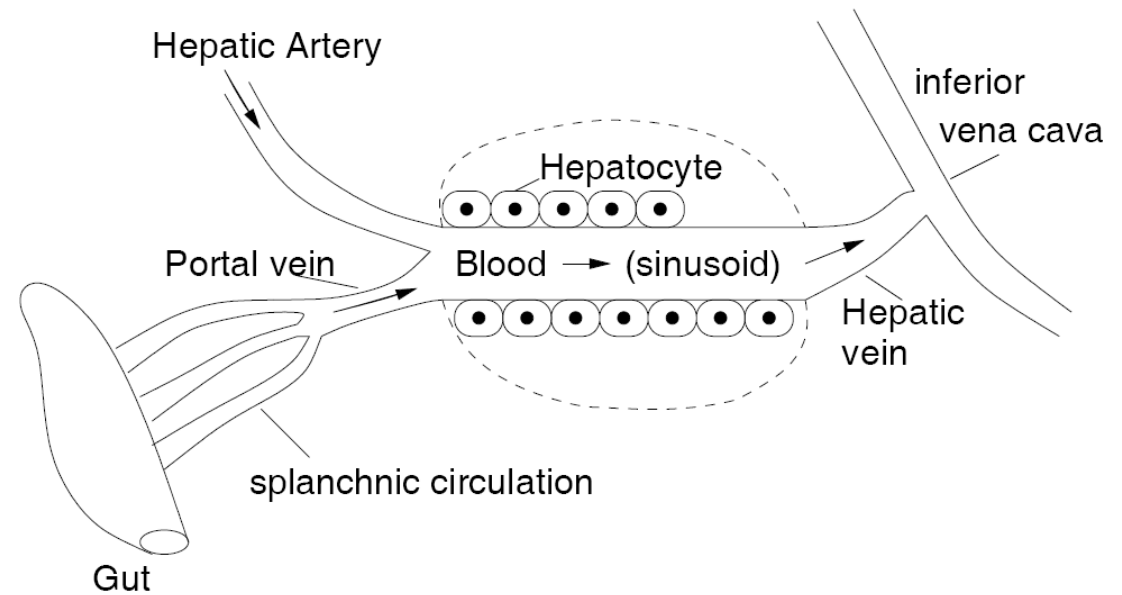
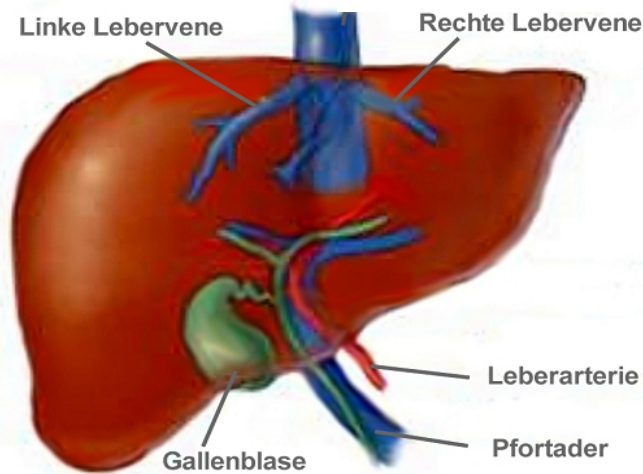
Approximate partition coefficients based on *in vitro* data



in vitro assay

$$P_{o:w} = \frac{C_{\text{octanol}}}{C_w} \approx \frac{C_{nl}}{C_{up}}$$

Hepatic metabolism



$$V_{tis} \frac{d}{dt} C_{tis} = Q_{tis} \cdot \left(C_{in} - \frac{C_{tis}}{K_{tis}} \right) - CL_{int} \cdot C_{tis}$$

intrinsic hepatic clearance
(potentially also saturable)

Determining hepatic intrinsic clearance from in vitro data

- Hepatocyte assay

$$CL_{\text{int}}(\textit{in vitro}) \text{ in } \left[\frac{\mu\text{L}}{\text{min}} \text{ per } 10^6 \text{ cells} \right]$$

- Microsomal assay

$$CL_{\text{int}}(\textit{in vitro}) \text{ in } \left[\frac{\mu\text{L}}{\text{min}} \text{ per } \mu\text{g} \text{ microsomes} \right]$$

- Scaling approach to hepatic intrinsic clearance

$$CL_{\text{int}}(\textit{in vivo}) = OW_{\text{liv}} \cdot SF \cdot CL_{\text{int}}(\textit{in vitro})$$

- with scaling factors

	hepato-cellularity	microsomal recovery
human	107-120 [10^6 hepatocytes/g liver]	33-52.5 [mg protein/g liver]
rat	109-135 [10^6 hepatocytes/g liver]	45-60 [mg protein/g liver]

Parameterization of PBPK models

Species specific

- blood flows, organ volumes
- tissue composition data

Drug specific

- intrinsic clearance CL_{int}
- blood:plasma ratio B:P
- fraction unbound f_u
- octanol-water coeff P_{ow}
- pKa value
- Administration (dose, route, etc)

Rate of change in tissue:

$$V_{tis} \frac{d}{dt} C_{tis} = Q_{tis} \cdot \left(C_{in} - \frac{C_{tis}}{K_{tis}} \right) - CL_{int} \cdot C_{tis}$$

Tissue-to-blood partition coefficient:

$$K_{tis} = \frac{C_{tis,ss}}{C_{blood,ss}}$$

Poulin/Theil (2000), Rodgers/Rowland (2005), Schmidt (2008)

Example

Example

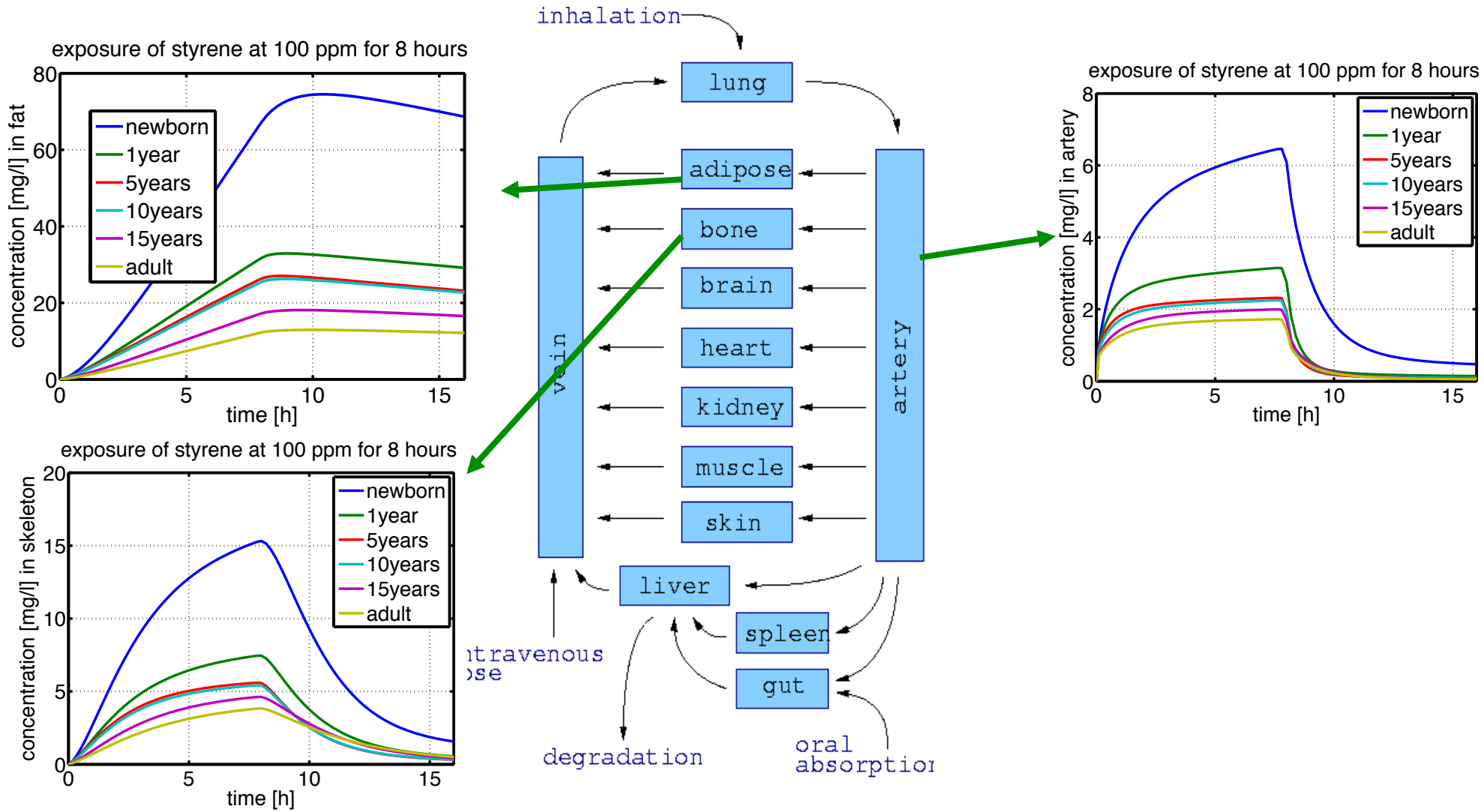
Arch Toxicol (2005) 79: 63–73
DOI 10.1007/s00204-004-0599-3

TOXICOKINETICS AND METABOLISM

Klaus Abraham · Hans Mielke · Wilhelm Huisinga
Ursula Gundert-Remy

Elevated internal exposure of children in simulated acute inhalation of volatile organic compounds: effects of concentration and duration

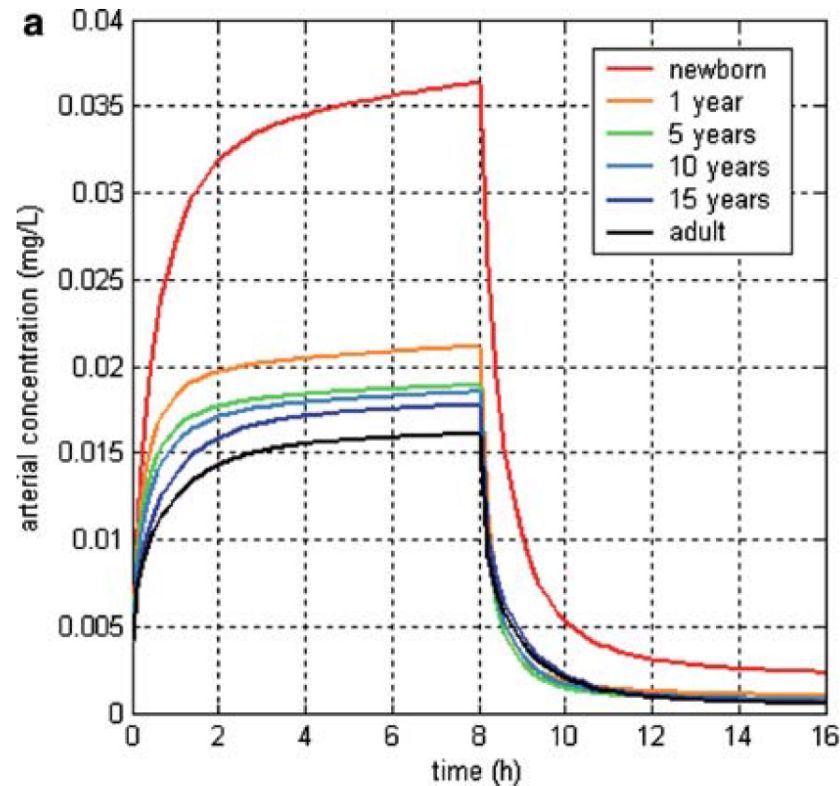
Example: exposure of styrene at 100 ppm for 8h



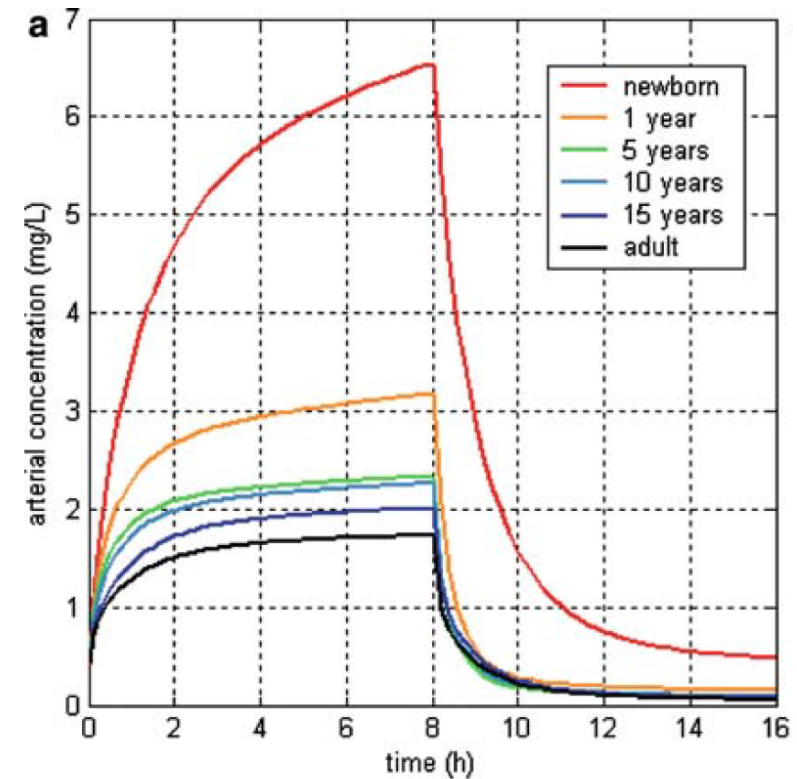
→ different concentrations and profiles in different tissues/organs

Non-linear relationship between external and systemic exposure

1 ppm

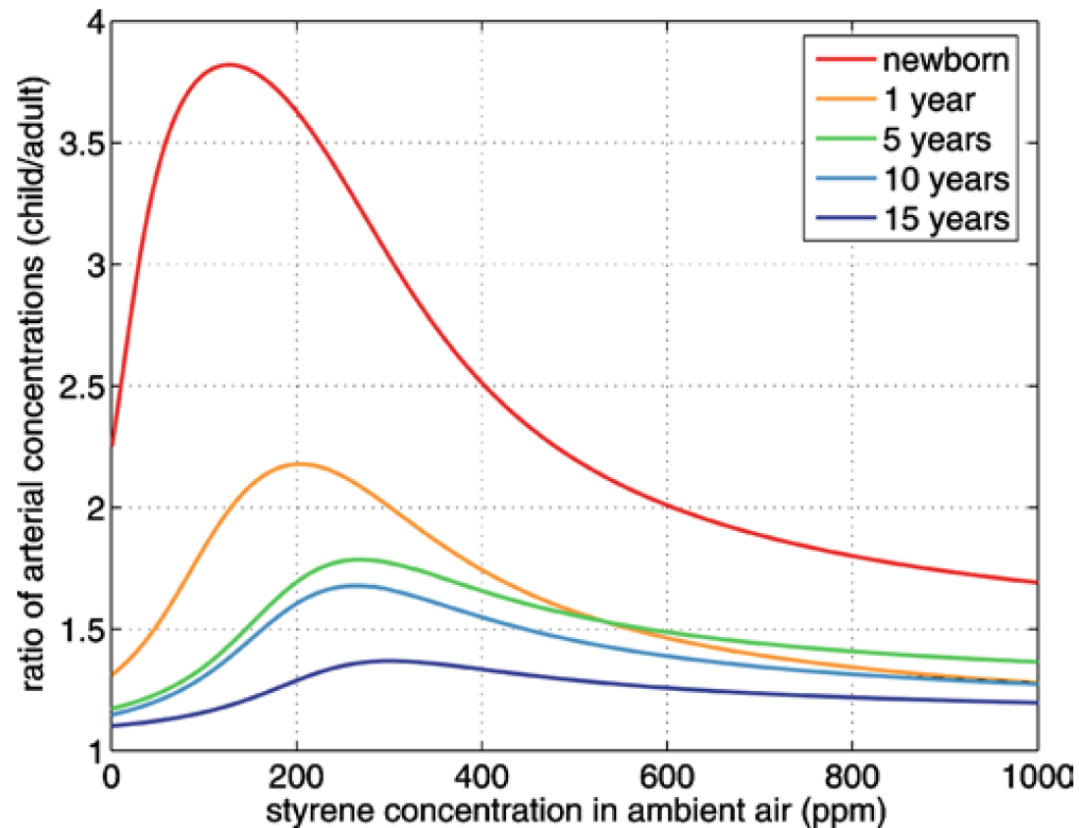


100 ppm



- Underlying reason: saturable metabolism

Exposure ratio Child:Adult depending on external exposure (at 8h)



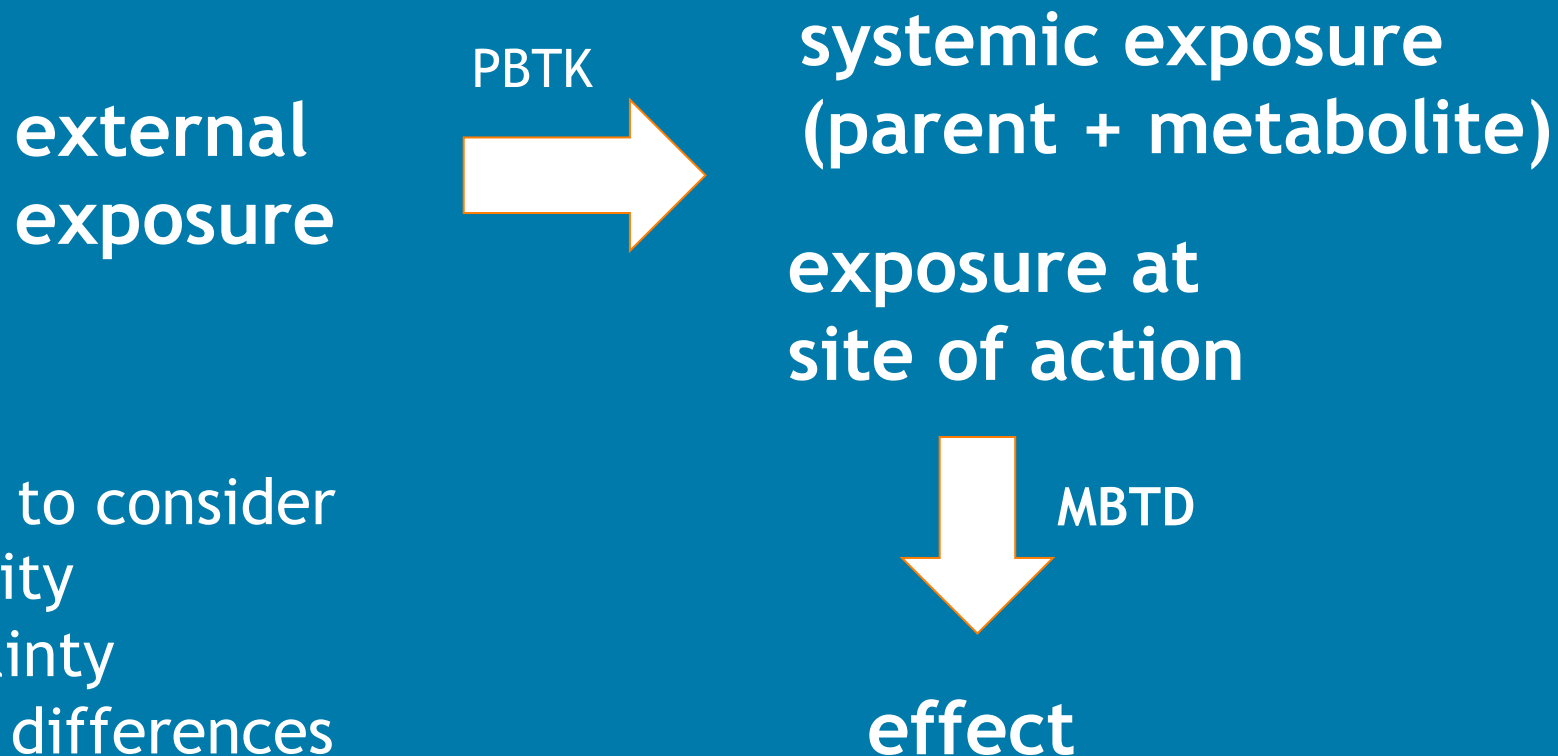
- exposure ratios also depend on time.
- Sensitivity analysis showed that predictions are most sensitive to alveolar ventilation rate with 20% change of $Q_p \rightarrow 14\%$ change in ratio(newborn)

Further examples

There are also a large number of good examples of PBPK models which describe the kinetics of important environmental contaminants, including methylene chloride (Andersen *et al.*, 1987a, 1991, 1994), trichloroethylene (Fisher *et al.*, 1991; Fisher and Allen, 1993; Allen and Fisher, 1993), chloroform (Corley *et al.*, 1990; Reitz *et al.*, 1990), 2-butoxyethanol (Johanson, 1986), kepone (Bungay *et al.*, 1981), polybrominated biphenyls (Tuey and Matthews, 1980), polychlorinated biphenyls (Lutz *et al.*, 1984) and dibenzofurans (King *et al.*, 1983), dioxins (Leung *et al.*, 1988; Kohn *et al.*, 1993; Andersen *et al.*, 1997), lead (O'Flaherty, 1991a,b,c,1993,1995b), arsenic (Mann *et al.*, 1996a,b), and methylmercury (Farris *et al.*, 1993).

Gentry, Clewell, Anderson, ENVIRON, manuscript

The potential



Important to consider

- variability
- uncertainty
- species differences
- in vitro-in vivo differences

Most critical bottleneck for application: availability of parameter values

Some references

- Gerlowski & Jain, *PBPK modelling: Principles and applications*, J Pharm Sci 72 (10), 1983
- Poulin & Krishnan/Theil, 1995-2002
- K. Abraham, H. Mielke, W. Huisinga and U. Gundert-Remy, *Elevated Internal Exposure of Children in Simulated Acute Inhalation of Volatile Organic Compounds: Effects of Concentration and Duration*, Arch Toxicol 79 (2005)
- M. Reddy et al, *Physiologically-based Pharmacokinetic Modeling*, Wiley 2005
- WHO, *Characterization and Application of PBPK models in Risk Assessment*, 2010
- H. Mielke, U. Gundert-Remy, *Physiologically Based Toxicokinetic Modelling as a Tool to Support Risk Assessment: Three Case Studies*, J Toxicology, 2012
- and many more.

Acknowledgement

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- For information:



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